

## Effects of Inhibitors of Ion-Motive ATPases on the Plasma Membrane Potential of Murine Erythroleukemia Cells

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**Summary.** The membrane electric effects of N,N'-dicyclohexylcarbodiimide (DCCD) and vanadate were studied in murine erythroleukemia cells (MELC), comparing the patch-clamp technique and the accumulation ratio ( $AR_{exp}$ ) of [<sup>3</sup>H]-tetraphenylphosphonium (TPP<sup>+</sup>). Electrophysiological measurements showed that both these inhibitors produce, at micromolar concentrations, a 20–30 mV hyperpolarization of resting potential ( $\Delta\psi_p$ ) of MELC, which is abolished when the electrochemical equilibrium potential of K<sup>+</sup> ( $E_K$ ) is brought close to zero.

DCCD and vanadate turned out to have distinct targets on the plasma membrane of MELC (an H<sup>+</sup> pump and the Na<sup>+</sup>,K<sup>+</sup>-ATPase, respectively).

Measurements of  $AR_{exp}$  showed that: (i) patch-clamp measurements of  $\Delta\psi_p$  were equivalent to those based on  $AR_{exp}$  of antimycin-pretreated cells ( $AR_{ANT}$ ); (ii) DCCD produced a strong increase in  $AR_{ANT}$ , that was antagonized by carbonyl cyanide *p*-trifluoromethoxyphenyl-hydrazone (FCCP) and diethylstilbestrol (DES); (iii) vanadate determined a marked increase in  $AR_{ANT}$  that was insensitive to FCCP, but antagonized by ouabain; (iv) incubation in high K<sup>+</sup> medium (HK) brought  $AR_{ANT}$  to 1.0 in the controls, but did not lower this ratio below 3.0 in the presence of DCCD or vanadate; (v) the total amount of TPP<sup>+</sup> taken up by the cells was in any case water extractable by a freezing and thawing procedure.

On the whole, our data indicate that DCCD and vanadate hyperpolarize the MELC by increasing the K<sup>+</sup> conductance and, at the same time, enhance the TPP<sup>+</sup> binding, probably by changing the electrostatic potential profile of the plasma membrane. These effects seem to involve functional modifications of the target pumps, apparently related to the ion-occluding state of these enzymes.

**Key Words** ion-motive ATPase inhibitors · leukemia cells · resting potential · potassium conductance · TPP<sup>+</sup> binding · electrostatic potential of the plasma membrane

### Introduction

Ion-motive ATPases (Pedersen & Carafoli, 1987a,b) mediate crucial cellular processes such as ATP synthesis or ion translocation against energy barriers.

In general, they are membrane-intrinsic proteins, composed of two functionally distinct moieties: a portion interacting with ATP in the cellular aqueous compartment (the catalytic portion) and the ion-translocating portion across the membrane (Amzel & Pedersen, 1983; Al-Awqati, 1986; Skou, 1988; Futai, Noumi & Maeda, 1989).

Various inhibitors greatly contributed to characterize diverse types of ion motive ATPases and to elucidate their functional properties (Glynn & Karlsh, 1975; Penefsky, 1979; Goffeau & Slayman, 1981; Pedersen & Carafoli, 1987a). In particular, DCCD specifically blocks the proton translocation step through the transmembrane moiety of H<sup>+</sup>-ATPases (Hoppe & Sebald, 1984), while vanadate predominantly interferes with the ion-translocating activity of Na<sup>+</sup>,K<sup>+</sup>-ATPases (Cantley, Resh & Guidotti, 1978; Glynn & Richards, 1982; Beaugé, 1988).

Interestingly, both DCCD (De La Peña et al., 1982; Baroni et al., 1988) and vanadate (Lichtshstein, Mullikin-Kilpatrick & Blume, 1982) have been reported to hyperpolarize the plasma membrane of many different types of cells, including yeasts and neuroblastoma cells. These studies have been performed essentially by measuring the intracellular accumulation of the lipophilic cation TPP<sup>+</sup>, but the hyperpolarizing properties of vanadate were confirmed by direct electrophysiological measurements. Although the hyperpolarization induced by DCCD in yeasts was traced to a K<sup>+</sup> efflux through K<sup>+</sup> channels (De La Peña et al., 1982; Baroni et al., 1988), while that sustained by vanadate was accompanied by an increase in the membrane conductance (Lichtshstein et al., 1982), the mechanism underlying these processes remains obscure so far.

We considered it worthwhile to explore this

topic by integrating the TPP<sup>+</sup> method and the patch-clamp technique, both of which have been employed widely on MELC (Arcangeli & Olivotto, 1986; Arcangeli et al., 1987*a,b*, 1989*a,b*). It was thought that this integration should provide information about the membrane energy profile, which might unravel the electric events, including the surface-charge effects, that occur in the membrane itself, when the ion-translocating moiety of a pump is blocked by specific inhibitors. In fact, the above-mentioned profile depends not only on the resting potential ( $\Delta\psi_p$ ), but also on the membrane dipole potential and on the diffuse double-layer surface potential (McLaughlin, 1977, 1989). While  $\Delta\psi_p$  can be exactly measured by the patch-clamp technique, accumulation of TPP<sup>+</sup> in cells and organelles is a function of both the resting potential and a complex of TPP<sup>+</sup> interactions with membranes, collectively referred to as "binding."

According to a widely accepted model, worked out on neutral phospholipid bilayers, binding consists of a confined adsorption of lipophilic ions to potential wells, formed near the membrane-water interfaces by the combination of repulsive electrostatic and attractive hydrophobic energy terms (Neumcke & Läuger, 1969; Ketterer, Neumcke & Läuger, 1971; Flewelling & Hubbell, 1986*a,b*). This model implies that lipid bilayers possess a dipole potential of several hundred millivolts positive in the interior, a feature of great importance to structural and functional properties of membrane proteins, including ion channels (Jordan, 1983). Variations of this potential, modifying the energy barrier in the middle of the membrane, may change the binding wells at the water interfaces. In natural membranes, containing negatively charged lipids, the latter generate a negative surface potential that represents another electrostatic parameter influenced (and mirrored) by the binding of lipophilic ions (Andersen et al., 1978; Bakker, 1982).

In this paper, we present patch-clamp and TPP<sup>+</sup> measurements aimed at clarifying the effects of DCCD and vanadate on both  $\Delta\psi_p$  and the membrane electrostatic potential profile in MELC. Electrophysiological data provided evidence that these inhibitors hyperpolarize the plasma membrane by increasing the K<sup>+</sup> conductance, while TPP<sup>+</sup> experiments indicated that this hyperpolarization is a direct consequence of the inhibitor interaction with the target pumps. Integration of data obtained with the two techniques allowed the conclusion that DCCD and vanadate cause a marked increase of TPP<sup>+</sup> binding, suggesting that changes of the electrostatic potential profile underlie the membrane electric effects of these inhibitors.

## Materials and Methods

### CELL CULTURE

MELC (strain 745A, originally obtained from Dr. Friend and kindly provided by Dr. G.B. Rossi, Istituto Superiore di Sanità, Rome) were routinely cultured in RPMI 1640 medium, supplemented with 5% foetal calf serum (FCS) ("complete standard medium"), as previously described (Arcangeli & Olivotto, 1986). Cells were harvested from preparatory cultures at  $1.8\text{--}2.0 \times 10^6$  cells/ml, centrifuged at  $250 \times g$  for 10 min, and, unless otherwise indicated, resuspended in freshly prepared complete medium at  $9 \times 10^5$  cells/ml. Usually 8 ml of this suspension were transferred into tissue culture flasks with a culture surface of 25 cm<sup>2</sup> and incubated at 37°C in a humidified atmosphere (5% CO<sub>2</sub> in air).

### ELECTROPHYSIOLOGICAL TECHNIQUES

$\Delta\psi_p$  was measured using an amplifier Axopatch 1-D (Axon Instruments), under the current-clamp configuration of the whole-cell patch-clamp technique (Hamill et al., 1981) on cells with an average diameter of about 8–9  $\mu\text{m}$ . All experiments were done at room temperature (20–22°C). In about 10% of the experiments a remarkable swelling of the cells occurred: these experiments were discarded on the assumption that the cells were not in good physiological condition. Glass pipettes (borosilicate, Hilgenberg, Germany), with a resistance of 5–10 M $\Omega$ , were filled with an internal solution adjusted at pH 7.3, containing (in mM): NaCl 14, KCl 120, MgCl<sub>2</sub> 2, 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid (HEPES) 10 and Na<sub>2</sub>ATP 3; the final Ca<sup>2+</sup> concentration was buffered with 10 mM ethylenebis (oxyethylenetri)-tetra acetic acid (EGTA) at  $10^{-7}$  M (*pCa* 7) or about  $10^{-9}$  M (*pCa* 9) by adding 4 mM or no CaCl<sub>2</sub>.

When needed, DCCD or Na-orthovanadate were added to these solutions, at the final concentrations of 5 and 25  $\mu\text{M}$ , respectively. External solutions were delivered with hypodermic needles inserted into a capillary with a small hole (inner diameter 0.4 mm) positioned in the vicinity of the cell under study. Unless otherwise indicated, the external solution was adjusted at pH 7.4 and contained (in mM): NaCl 140, KCl 3, MgCl<sub>2</sub> 2, CaCl<sub>2</sub> 2, glucose 5 and HEPES 10. Modifications of the electrochemical equilibrium potential of K<sup>+</sup> ( $E_K$ ) were obtained by changing the concentrations of KCl, either in the external or in the internal solution, and adjusting accordingly the NaCl concentration in order to keep constant the osmolarity. All seals were over 10 G $\Omega$ . The plasma membrane resistance was measured by injecting a series of 5-pA impulses (10 Hz) and recording the subsequent shifts of  $\Delta\psi_p$  (Fig. 1D). Analog signals, filtered at 1 kHz, were recorded on a video tape, through a PCM processor (Sony, model 701ES) and analyzed off-line using the pCLAMP hardware-software system (Axon Instruments).

### MEASUREMENT OF ACID EXTRUSION

Acid efflux from the cells was detected by continuously monitoring the extracellular pH ( $\text{pH}_e$ ) with a conventional combination pH electrode, in a water-jacketed chamber at 37°C, containing 1.5 ml of an unbuffered saline of the following composition (in mM): NaCl 143, KCl 3, CaCl<sub>2</sub> 2 and MgCl<sub>2</sub> 2. Acid loading of cells was performed essentially according to Swallow, Grinstein & Rotstein, (1988). Cells ( $4\text{--}6 \times 10^6$ ) were incubated in 2 ml of

RPMI 1640 medium, containing 10 mM HEPES and 40 mM  $\text{NH}_4\text{Cl}$  for 15 min at 37°C. After this incubation, cells were centrifuged for 5 min at  $200 \times g$ , gently resuspended in the unbuffered saline and immediately transferred into the chamber. Cells were magnetically stirred and the  $\text{pH}_e$  monitored for about 10 min. Initial  $\text{pH}_e$  at the moment of cell resuspension in the saline varied between 7.50 and 7.70.

#### DETERMINATION OF INTRACELLULAR ATP

Intracellular ATP was determined by high pressure liquid chromatography (HPLC), according to Pogolotti and Santi (1982). Preparation of cell extracts was as follows: cells were seeded in complete medium at  $9 \times 10^5/\text{ml}$  in 75-cm<sup>2</sup> tissue culture flasks and incubated for 1 hr at 37°C. At the end of this incubation cells were harvested, centrifuged at  $250 \times g$  at 4°C for 5 min and the pellet extracted with 1 M perchloric acid at 2°C. Acid extract was frozen and thawed twice in liquid nitrogen, maintained for 30 min in ice and then centrifuged at 13,000 rpm for 20 min at 4°C in a Heraeus centrifuge (Biofuge 17 RS). The supernatant was collected, buffered with Tris HCl 1 M + KOH 2 N, and then centrifuged at 3,000 rpm for 5 min in a Heraeus Biofuge A centrifuge. The supernatant was filtered and used for ATP determination by means of a HPLC apparatus (Waters).

#### DETERMINATION OF INTRACELLULAR WATER AND INTRA- AND EXTRACELLULAR ION CONCENTRATIONS

Intracellular water ( $\text{H}_2\text{O}_i$ ),  $\text{Na}^+$  and  $\text{K}^+$  concentrations in intra- ( $[\text{Na}^+]_i$ ,  $[\text{K}^+]_i$ ), and extracellular ( $[\text{Na}^+]_e$ ,  $[\text{K}^+]_e$ ) media were determined as previously described (Arcangeli & Olivotto, 1986; Arcangeli et al., 1987a).

#### DETERMINATION OF $\text{TPP}^+$ ACCUMULATION RATIO ( $AR_{\text{exp}}$ )

This ratio was determined as previously described (Arcangeli & Olivotto, 1986).

#### THE ESTIMATE OF WATER-EXTRACTABLE $\text{TPP}^+$

Water-extractable  $\text{TPP}^+$  was determined by labeling MELC with  $^3\text{H}\text{-TPP}^+$  as previously described (Arcangeli & Olivotto, 1986). At the end of the labeling, cells were centrifuged at  $250 \times g$  for 10 min and the pellet was resuspended in 250  $\mu\text{l}$  of distilled water. Cells were immediately frozen and thawed twice, then centrifuged in a Beckman Microfuge at  $12,000 \times g$  for 5 min. Radioactivity in the supernatant was measured in a Packard Tri Carb 460 D scintillation counter. The percentage of water-extractable  $\text{TPP}^+$  ( $\text{TPP}_{\text{we}}^+$ ) was calculated by dividing the dpm contained in the supernatant, obtained as above, by the total dpm extracted from the cells as previously described (Arcangeli & Olivotto, 1986).

#### THE RATIONALE OF $\text{TPP}^+$ MEASUREMENTS

Following previous contributions to this topic (Ketterer et al., 1971; McLaughlin, 1977; Rottenberg, 1979; Felber & Brand, 1982; Azzone, Pietrobon & Zoratti, 1984; Ritchie, 1984; Flewelling &

Hubbell, 1986a,b), we based the use of  $\text{TPP}^+$  on the following rationale.

In view of the scanty endoplasmic reticulum of MELC (Arcangeli & Olivotto, 1986), we have assumed that mitochondria are the only cellular organelles whose membrane potential could affect the amount of  $\text{TPP}^+$  taken up by the cells. Hence

$$AR_{\text{exp}} = \frac{\text{TPP}_i^+}{\text{H}_2\text{O}_i} \cdot \frac{1}{[\text{TPP}^+]_e} = 1 + C\Delta\psi_p + C\Delta\psi_m + C_B \quad (1a)$$

where  $AR_{\text{exp}}$  represents the  $\text{TPP}^+$  accumulation ratio between the intracellular and the extracellular compartments;  $\text{TPP}_i^+$  is the total  $\text{TPP}^+$  taken up by the cells;  $\text{H}_2\text{O}_i$  is the intracellular water;  $[\text{TPP}^+]_e$  is the  $\text{TPP}^+$  concentration in the bulk extracellular water;  $C\Delta\psi_p$  is the  $AR$  value accounted for by the  $\Delta\psi_p$  contribution to  $AR_{\text{exp}}$ ;  $C\Delta\psi_m$  is the  $AR$  value accounted for by the contribution to  $AR_{\text{exp}}$  of the mitochondrial resting potential ( $\Delta\psi_m$ );  $C_B$  is the  $AR$  value accounted for by the contribution to  $AR_{\text{exp}}$  of the binding (see Introduction) and the term 1 is the value assumed by  $AR_{\text{exp}}$  when all the above contributions are null, so that at the equilibrium  $\frac{\text{TPP}_i^+}{\text{H}_2\text{O}_i} = [\text{TPP}^-]_i = [\text{TPP}^+]_e$  where  $[\text{TPP}^+]_i$  is the  $\text{TPP}^+$  concentration in the total cellular water.

At the equilibrium partitioning, the  $\text{TPP}^+$  accumulation ratio between the extracellular and the extramitochondrial cell compartments is governed by  $\Delta\psi_p$  according to the Nernst equation, as follows

$$\Delta\psi_p \text{TPP}^+ = -61.5 \log \left( \frac{[\text{TPP}^+]_{\text{em}}}{[\text{TPP}^+]_e} = AR_p \right) \quad (2a)$$

where  $[\text{TPP}^+]_{\text{em}}$  is the  $\text{TPP}^+$  concentration in the extramitochondrial compartment and  $-61.5 = 2.3 \cdot RT/F$ .

By analogy, the accumulation ratio between the cell intra- and extramitochondrial compartments is governed by the mitochondrial potential ( $\Delta\psi_m$ ) as follows

$$\Delta\psi_m \text{TPP}^- = -61.5 \log \left( \frac{[\text{TPP}^-]_{\text{im}}}{[\text{TPP}^-]_{\text{em}}} = AR_m \right) \quad (3)$$

where  $[\text{TPP}^-]_{\text{im}}$  is the  $\text{TPP}^+$  concentration in the intramitochondrial compartment.

According to Eq. (2a), when  $\Delta\psi_p$  is zero,  $AR_p = 1$ . Under these conditions  $C\Delta\psi_p$  of Eq. (1a) is zero; on the other hand, when  $\Delta\psi_p$  is different from zero, its contribution to  $AR_{\text{exp}}$  will depend both on  $AR_p$  and on the relative volume of the extramitochondrial compartment as compared to the cell total volume. Thus, on the whole

$$C\Delta\psi_p = (AR_p - 1) \cdot (S_t - S_m)/S_t \quad (4a)$$

where  $S_t$  and  $S_m$  are the cell total volume and the volume of the total mitochondrial compartment.

By analogy

$$C\Delta\psi_m = (AR_m - 1) \cdot S_m/S_t \quad (5a)$$

Recalling Eqs. (2a) and (3), one can express  $AR_m$  as a function of  $[\text{TPP}^+]_e$ , so that Eq. (5a) becomes

$$C_{\Delta\psi_m} = \frac{[\text{TPP}^+]_{\text{im}}}{AR_p [\text{TPP}^+]_e} - 1 \cdot S_m/S_i \quad (5b)$$

which expresses how in Eq. (1a)  $AR_{\text{exp}}$  depends on  $\Delta\psi_m$ .

In MELC,  $S_m = 0.039 S_i$ , or  $(S_i - S_m)/S_i = 0.96$  (Arcangeli & Olivotto, 1986). Hence Eq. (4a) becomes

$$C_{\Delta\psi_p} \approx AR_p - 1. \quad (4b)$$

Hence, whenever  $C_B = 0$ , and provided that  $\Delta\psi_m$  is abolished by the mitochondrial inhibitor antimycin A (ANT) (Nicholls, 1982), it follows that

$$AR_{\text{ANT}} = 1 + C_{\Delta\psi_p} \approx AR_p \quad (1b)$$

and, substituting in Eq. (2a)

$$\Delta\psi_p \text{TPP}^+ = -61.5 \log AR_{\text{ANT}}. \quad (2b)$$

On the other hand, whenever  $C_B$  is not negligible

$$AR_{\text{ANT}} = 1 + C_{\Delta\psi_p} + C_B. \quad (1c)$$

In cells, where, like in MELC,  $\Delta\psi_p$  is mainly governed by the  $K^+$  conductance (Arcangeli et al., 1987a,b, 1989b),  $C_{\Delta\psi_p}$  can be brought close to zero when  $E_K$  is approximated to zero by incubating the cells in high  $K^+$  external medium (HK). Then, in this medium and in ANT, one has

$$AR_{\text{HK+ANT}} = 1 + C_B \text{HK} \quad (1d)$$

which gives an estimate of the binding in HK.

## MATERIALS

Tissue culture flasks were purchased from Sterilin; the petri dishes used in patch-clamp experiments were from Falcon. RPMI 1640 medium was obtained from GIBCO Laboratories; FCS was from Flow Laboratories. DCCD, Na-orthovanadate, DES and FCCP were purchased from Aldrich; antimycin A and ouabain were from Sigma. Oligomycin and ATP were obtained from Boehringer Mannheim. Amiloride was a kind gift of Merk Sharp and Dohme.  $^3\text{H-TPP}^+$  (bromide salt, sp act 24 Ci/mmol) was obtained from Amersham Radiochemical Centre and diluted with cold  $\text{TPP}^+$  provided by ICN, K and K Laboratories. All other reagents were of analytical grade.

$\text{TPP}^+$ , DCCD, DES, FCCP, antimycin A and oligomycin were dissolved in ethanol and diluted in the selected media immediately before use. At the corresponding dilutions, ethanol was always ineffective on the parameters under study.

## Results and Discussion

### THE PATCH-CLAMP DEMONSTRATION OF THE $\Delta\psi_p$ HYPERPOLARIZING EFFECTS OF DCCD AND VANADATE

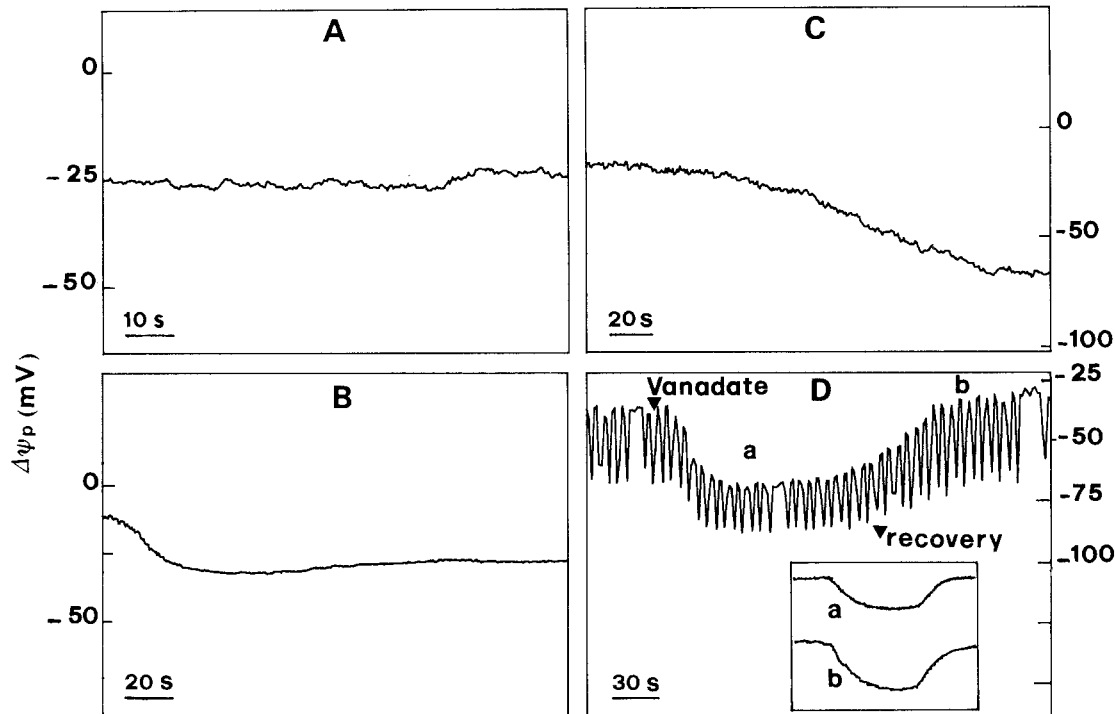
The effects of ATPase inhibitors on  $\Delta\psi_p$  were first analyzed by a series of patch-clamp experiments, whereby the inhibitors were added to the cells from

the inside. This allowed the use of very low inhibitor concentrations and ruled out possible hyperpolarizing effects caused by vanadate anion diffusion from the outside into the cells (Lichtshtein et al., 1982).

Figure 1 shows typical whole-cell recordings of  $\Delta\psi_p$ , either in the absence or in the presence of the ATPase inhibitors. Using the control pipette solution,  $\Delta\psi_p$  did not undergo any significant variation during the measuring time (Fig. 1A). When the pipette solution contained 5  $\mu\text{M}$  DCCD (Fig. 1B) or 25  $\mu\text{M}$  vanadate (Fig. 1C), a substantial hyperpolarization of  $\Delta\psi_p$  occurred constantly, although at variable times after the beginning of the measurement. Table 1,A shows that, at the steady state, both DCCD and vanadate produce, in physiological media, an average significant hyperpolarization (20 mV), with respect to the controls. When  $E_K$  was lowered from  $-98$  to  $-10$  mV by increasing  $[\text{K}^+]_e$  in the perfusion solutions,  $\Delta\psi_p$  followed  $E_K$  either in the absence or in the presence of the inhibitors (Table 1,A). The same occurred when  $E_K$  was changed to  $-18$  mV, essentially by lowering the  $K^+$  concentration in the internal solution (Table 1,B). The fact that in the controls  $\Delta\psi_p$  depends on  $E_K$  indicates that a marginal potassium conductance is present under these conditions. Thus, under the simplifying assumption of ohmic relations, the resting equilibrium condition gives  $g_K(\Delta\psi_p - E_K) = -g_L(\Delta\psi_p - E_L)$ , where  $g_K$ ,  $E_K$ ,  $g_L$  and  $E_L$  are the conductances and Nernst potentials for potassium and the leakage pathways, respectively. The ratio  $g_K/g_L = -(\Delta\psi_p - E_L)/(\Delta\psi_p - E_K)$  is reported in Table 1,A either for starting ( $g_{K_i}/g_L$ ) or steady-state ( $g_{K_f}/g_L$ ) conditions, assuming  $E_K = -98$  and  $E_L = 10$  (Arcangeli et al., 1989b). Notice that all the  $g_{K_i}/g_L$  values were not significantly different from the  $g_{K_f}/g_L$  value of the control. On the contrary,  $g_{K_f}/g_L$  increased in the presence of inhibitors. A comparative index of this effect is reported in Table 1,A as  $g_{K_f}/g_{K_i}$ , which shows that the  $K^+$  conductance is increased more than twofold by both inhibitors.

On the whole, data in Table 1 demonstrate that the hyperpolarization induced by DCCD and vanadate in physiological media depends on the increase in  $K^+$  conductance, in keeping with the indirect indications previously reported in yeasts (De la Peña et al., 1982; Baroni et al., 1988).

This increase most likely accounts for the report that, when added to the external medium, vanadate produces a  $\Delta\psi_p$  hyperpolarization associated to a diminution of the electrical resistance of the plasma membrane (Lichtshtein et al., 1982). We confirmed this finding. In fact 5 mM extracellular vanadate determined a 30–40 mV hyperpolarization of  $\Delta\psi_p$  (Fig. 1D), associated to an increase in the membrane conductance, as revealed by the response to current



**Fig. 1.** Patch-clamp recordings of DCCD and vanadate effects on  $\Delta\psi_p$  and on the plasma membrane conductance. (A–D) =  $\Delta\psi_p$  records obtained in the whole-cell configuration as indicated in Materials and Methods. (A) control.  $\Delta\psi_p$  does not vary significantly during the recording time. (B) The pipette solution contained  $5 \mu\text{M}$  DCCD. The  $\Delta\psi_p$  hyperpolarizes within the first minute and reaches a steady-state value more negative than the initial one. (C) The pipette solution contained  $25 \mu\text{M}$  vanadate.  $\Delta\psi_p$  shows a progressive hyperpolarization followed by a steady-state value. (D) Measurement of the plasma membrane conductance in the absence and in the presence of vanadate. The pipette contained the standard control saline, while  $5 \text{ mM}$  vanadate was added to the external solution at the time indicated by the first arrow. The membrane conductance was measured by injecting low frequency (10 Hz) 5-pA current pulses as reported in Materials and Methods. On the average, the membrane conductance was  $6.67 \text{ G}\Omega$  ( $\pm 0.441 \text{ SD}$ ;  $n = 15$ ), corresponding to  $0.15 \text{ nS}$ , in the absence, and  $3.85 \text{ G}\Omega$  ( $\pm 0.181 \text{ SD}$ ;  $n = 10$ ), corresponding to  $0.26 \text{ nS}$ , in the presence of vanadate. After recovery, both steady state  $\Delta\psi_p$  and conductance return to the initial values. Under the conditions used in all these experiments, electrochemical equilibria were as follows (in mV):  $E_{\text{K}} = -104$ ;  $E_{\text{Na}} = +56$ ;  $E_{\text{Cl}} = 0$ . Inset: The hyperpolarizing response to one current pulse during (trace a) and after (trace b) vanadate perfusion.

pulses injected into the cell (see inset in Fig. 1D). The simple application of the procedure used for Table 1,A gave  $g_{\text{Kf}}/g_{\text{Ki}} = 3.53$ . With the help of the voltage response shown in Fig. 1D, we have calculated  $g_{\text{Ki}} + g_{\text{L}} = 0.15 \text{ nS}$  and  $g_{\text{Kf}} + g_{\text{L}} = 0.26 \text{ nS}$ . This gave us, in this particular example, a direct quantification of  $g_{\text{Ki}} = 0.061 \text{ nS}$  and  $g_{\text{Kf}} = 0.185 \text{ nS}$ , or  $g_{\text{Kf}}/g_{\text{Ki}} = 3.04$ . This value is not far from the previous 3.53, supporting the validity of the method worked out to measure the relative  $g_{\text{K}}$  increase in Table 1,A.

Since, in MELC,  $\Delta\psi_p$  is mainly regulated by  $\text{Ca}^{2+}$ -dependent  $\text{K}^+$  channels ( $K_{\text{Ca}}$ ) (Arcangeli et al., 1987a,b, 1989b), the hyperpolarization produced by ATPase inhibitors could be reasonably traced back to a  $[\text{Ca}^{2+}]_i$  increase, consequent to the block of  $\text{Ca}^{2+}$ -ATPases (Pedersen & Carafoli, 1987a). However, this explanation does not easily apply to our data, since the hyperpolarizing activity of DCCD

and vanadate was maintained when  $\text{Ca}^{2+}$  in the pipette was lowered below  $10^{-9} \text{ M}$ , a concentration usually incompatible with the activation of these channels (Hamill, 1983). The involvement of other types of  $\text{K}^+$  channels, such as those recently identified as being inhibited by ATP (Ashcroft, Harrison & Ashcroft, 1984), is equally difficult to conceive, since  $3 \text{ mM}$  ATP was present in the pipette solution used in these experiments.

#### CELL PARAMETERS AFFECTED BY THE ATPASE INHIBITORS AND IDENTIFICATION OF THE TARGET PUMPS

A deeper insight into the mechanism of the hyperpolarization sustained by DCCD and vanadate was sought by analyzing the alteration induced in intact cells and, hence, trying to identify the ion-motive ATPases principally affected.

**Table 1.** Effects of DCCD and vanadate on  $\Delta\psi_p$  measured by the patch-clamp technique at the beginning or at the steady state of measurement

	Start		Steady state			
	$\Delta\psi_p$ (mV)	$g_{K_i}/g_L$	$\Delta\psi_p$ (mV)	LK ( $E_K = -98$ ) $g_{K_f}/g_L$	$g_{K_f}/g_{K_i}$	HK ( $E_K = -10$ ) $\Delta\psi_p$ (mV)
A						
Control	-24.0 $\pm 3.5$ (7)	0.48 $\pm 0.07$ (7)	-17.4 $\pm 2.8$ (7)	0.35 $\pm 0.04$ (7)	0.75 $\pm 0.05$ (7)	-3.7 $\pm 2.9$ (4)
DCCD 5 $\mu\text{M}$	-21.0 $\pm 5.5$ (3)	0.42 $\pm 0.1$ (3)	-38.6* $\pm 1.2$ (5)	0.82 $\pm 0.04$ (5)	2.18 $\pm 0.37$ (3)	-6.0 $\pm 3.6$ (4)
Vanadate 25 $\mu\text{M}$	-18.7 $\pm 3.7$ (6)	0.38 $\pm 0.07$ (6)	-37.5** $\pm 6.1$ (6)	1.08 $\pm 0.29$ (6)	2.66 $\pm 0.86$ (6)	-8.0 $\pm 5.2$ (4)
B			( $E_K = -18$ )			
Control	-23.5 $\pm 7.17$ (12)		-9.7 $\pm 3.87$ (12)			
DCCD 5 $\mu\text{M}$	-20 $\pm 5.62$ (6)		-10.2 $\pm 5.46$ (6)			
Vanadate 25 $\mu\text{M}$	-19.8 $\pm 7.14$ (8)		-13.3 $\pm 5.25$ (8)			

\* $P < 0.001$ ; \*\* $P < 0.01$  (Student's  $t$  test for independent samples).

Experimental conditions were as reported in the legend to Fig. 1 A–C. *Start* refers to  $\Delta\psi_p$  values recorded soon after the whole-cell configuration was attained, *steady state* to the plateau subsequently attained and maintained for at least 2 min.  $g_{K_i}$  and  $g_{K_f}$  refers to  $g_K$  calculated at the start or at the steady state, respectively. Values are means  $\pm$  SEM of the number of experiments listed in parentheses.

A:  $E_K$  was varied from  $-98$  to  $-10$  mV, passing from a perfusing solution with 3 mM  $K^+$  (low  $K^+ = LK$ ) to another with 80 mM  $K^+$  (high  $K^+ = HK$ ). Osmolarity was compensated by changing  $[Na^+]_e$ .  $[K^+]_i = 120$  mM. For calculation of  $g_K/g_L$  and  $1 + g_{K_f}/g_{K_i}$ , see text.

B:  $[K^+]_i$  was 20 mM and  $[K^+]_e$  10 mM. By these means,  $E_K$  was maintained constantly at  $-18$  mV.

As shown in Table 2, the rate of  $H^+$  extrusion from acid-loaded MELC is strongly reduced by DCCD, as well as by another classical  $H^+$ -ATPase inhibitor, DES. Although these data support the existence of a plasma membrane  $H^+$  pump, the activity of the latter is not the only mechanism of proton extrusion from MELC, since a marked inhibition of this parameter is produced also by a specific inhibitor of the  $Na^+/H^+$  antiport, such as amiloride (Benos, 1982). This antiport could explain the finding, shown in Table 3, that DCCD, although ineffective on  $[K^+]_i$  and  $(H_2O)_i$ , increases  $[Na^+]_i$ . The lack of effect on  $[K^+]_i$  clearly indicates that no inhibition of the  $Na^+, K^+$ -ATPase is brought about by DCCD up to 200  $\mu\text{M}$ . On the other hand, the  $[Na^+]_i$  increase probably reflects an activation of the  $Na^+/H^+$  antiport following the diminution of the proton pump activity.

Contrary to DCCD, vanadate is a potent inhibitor of the MELC  $Na^+, K^+$ -ATPase, causing a dose-de-

pendent decrease in  $[K^+]_i$  and an increase in  $[Na^+]_i$  (Table 3). At the concentration of 5 mM, vanadate completely mimics the effects of ouabain, namely the most specific inhibitor of this pump (Wallick & Shwartz, 1988).

On the whole, we conclude that DCCD and vanadate differ substantially as to their preferential target ATPase on the MELC plasma membrane, which are a  $H^+$  pump and the  $Na^+, K^+$ -ATPase, respectively. On the contrary, as shown in Table 4, both 50  $\mu\text{M}$  DCCD and 1 mM vanadate produce only a modest decrease in intracellular ATP content, similar to that produced by oligomycin at 4  $\mu\text{g}/10^7$  cells. Since, as indicated by the drastic effect of antimycin, the mitochondrial contribution to this parameter is decisive, the data in Table 4 suggest that, at the tested concentrations, DCCD and vanadate behave similarly to oligomycin at nonsaturating doses, with respect to their inhibitory activity on the mitochondrial  $F_0-F_1$ -ATPase.

**Table 2.** Effect of various inhibitors on H<sup>+</sup> efflux rate in acid-loaded MELC

	H <sup>+</sup> extrusion rate (pmol/min/10 <sup>6</sup> cells)
Control	2.71 ± 0.16 (5)
DCCD 50 μM	1.30 ± 0.07 (2)*
DCCD 200 μM	1.40 ± 0.09 (7)*
Amiloride 200 μM	1.39 ± 0.09 (2)*
DES 50 μM	1.27 ± 0.18 (3)*

\*  $P < 0.001$  (Student's  $t$  test for independent samples).

Cells were resuspended at  $2.5 \times 10^6$ /ml in 2 ml of a modified RPMI medium, buffered with 10 mM HEPES and containing 40 mM NH<sub>4</sub>Cl. When needed, DCCD was added to the cell suspension at the concentrations indicated. After 15 min of incubation at 37°C, cells were centrifuged and gently resuspended in 1.5 ml of the unbuffered saline (see Materials and Methods) and transferred into the jacketed chamber at 37°C. pH<sub>e</sub> was continuously recorded and the initial rate of acid extrusion measured as described in Materials and Methods. Amiloride and DES were added into the chamber after the beginning of pH<sub>e</sub> recording. Values are means ± SEM of the number of experiments listed in parentheses.

#### THE EQUIVALENCE OF TPP<sup>+</sup> AND PATCH-CLAMP MEASUREMENTS OF Δψ<sub>p</sub> UNDER THE STANDARD CONDITIONS

In Table 5, the electrophysiological measurements of Δψ<sub>p</sub> obtained with the whole-cell configuration in the control cells are compared to values of Δψ<sub>p</sub> TPP<sup>+</sup> calculated in antimycin-pretreated cells, according to Eq. (2b). As shown, the two types of measurements agree very well over the range of Δψ<sub>p</sub> variations induced in MELC by changing the cell inoculum (Arcangeli & Olivotto, 1986; Arcangeli et al., 1989a). Furthermore, whatever the method of measurement used, Δψ<sub>p</sub> of leukemia cells follows the variations of E<sub>K</sub> produced by the increase of [K<sup>+</sup>]<sub>e</sub>, in keeping with the prominent role of the K<sub>Ca</sub> chan-

nels in the regulation of this potential. While confirming the validity of Eq. (2b) under the standard conditions of culture, these results indicate that all the parameters, except  $C\Delta\psi_p$ , identified in Eq. (1a) as potentially affecting  $AR_{exp}$ , are negligible under these conditions. This means that: (i) the contribution of Δψ<sub>m</sub> to the cell uptake of TPP<sup>+</sup> is abolished by antimycin; and (ii) the TPP<sup>+</sup> binding is negligible, confirming our previous findings (Arcangeli & Olivotto, 1986). In turn, the equivalence of the patch-clamp measurements with those obtained by means of the TPP<sup>+</sup> technique, which leaves the internal cellular milieu unaltered, indicates that our electrophysiological records are not biased by the unavoidable alterations of this milieu in the whole-cell configuration.

#### THE EFFECTS OF ATPASE INHIBITORS ON AR<sub>exp</sub>

Basing on the above, we performed the following analysis of the effects of ATPase inhibitors on AR<sub>exp</sub> (Table 6).

#### THE EFFECTS OF OLIGOMYCIN

AR<sub>exp</sub> is not significantly affected in the controls either by inhibitors of the respiratory chain (antimycin) or by protonophores (FCCP). This behavior indicates that the contribution of Δψ<sub>m</sub> to AR<sub>exp</sub> ( $C\Delta\psi_m$ ) is negligible, without necessarily implying that Δψ<sub>m</sub> is null. In fact, taking into account the small percentage of cell total volume occupied by mitochondria (3.9%), it can be calculated that in MELC, up to Δψ<sub>m</sub> values of -60 mV,  $C\Delta\psi_m$  remains within the SE limits of AR<sub>exp</sub>. However, these low values of Δψ<sub>m</sub> contrast with the marked mitochondrial contribution to cellular ATP, witnessed by the fall of this parameter produced by antimycin treat-

**Table 3.** Effects of various ATPase inhibitors on [Na<sup>+</sup>]<sub>i</sub>, [K<sup>+</sup>]<sub>i</sub> and H<sub>2</sub>O<sub>i</sub>

	[K <sup>+</sup> ] <sub>i</sub> (mM)	[Na <sup>+</sup> ] <sub>i</sub> (mM)	H <sub>2</sub> O <sub>i</sub> (nl/10 <sup>6</sup> cells)
Control	152.3 ± 9.9 (8)	14.1 ± 2.0 (8)	343.6 ± 26.9 (8)
DCCD 50 μM	151.5 ± 12.3 (5)	35.9 ± 6.6 (5)	364.6 ± 39.9 (5)
DCCD 200 μM	159.5 ± 13.5 (2)	51.8 ± 8.8 (2)	346.5 ± 68.5 (2)
Vanadate 1 mM	81.0 ± 12.0 (2)	63.0 ± 12.0 (2)	294.5 ± 14.8 (2)
Vanadate 5 mM	20.0 ± 3.0 (2)	135.5 ± 18.5 (2)	346.5 ± 27.5 (2)
Ouabain 6 mM	32.4 ± 3.0 (7)	138.3 ± 14.0 (7)	398.0 ± 46.0 (8)

Cells were seeded at  $9 \times 10^5$ /ml in complete standard medium for 90 min. Inhibitors, at the concentrations indicated, were added 1 hr after cell seeding, except ouabain, that was added 30 min after seeding. At the end of incubation, cells were processed as described in Materials and Methods. Values are means ± SEM of the number of experiments listed in parentheses.

**Table 4.** Effects of various inhibitors on intracellular ATP content

	ATP (nmol/10 <sup>6</sup> cells)
Control	3.40 ± 0.35 (12)
Antimycin A	1.17 ± 0.24 (8)
Oligomycin (4 μg)	2.54 ± 0.34 (6)
Oligomycin (40 μg)	1.74 ± 0.28 (7)
DCCD 50 μM	2.78 ± 0.43 (6)
Vanadate 1 mM	2.86 ± 0.75 (3)
Vanadate 5 mM	1.26 ± 0.75 (2)

Cells were seeded into 75-cm<sup>2</sup> tissue culture flasks and incubated in complete standard medium for 90 min. Antimycin A (20 μM) was added 30 min after seeding; oligomycin (4 or 40 μg/10<sup>7</sup> cells), DCCD (50 μM) and vanadate (1 and 5 mM) were added 1 hr after cell seeding. At the end of incubation, cells were processed for ATP determination as indicated in Materials and Methods. Values are means ± SEM of the number of experiments listed in parentheses.

ment (Table 4). This seems to rule out the possibility that the low  $\Delta\psi_m$  of leukemic cells reflects a deenergized state of mitochondria, a conclusion confirmed by the effects of a specific inhibitor of  $F_0F_1$ -ATPase, such as oligomycin. In fact, as shown in Table 6 and Fig. 2, this inhibitor caused a strong increase of  $AR_{exp}$  that was abolished by cell pretreatment with either antimycin or FCCP. These data are consistent with the predictions of the chemiosmotic theory (Mitchell, 1961) for tightly coupled mitochondria, whereby the block of  $F_0F_1$ -ATPase interrupts

the drop of potential required to sustain the  $H^+$  current across this enzyme; this produces the maximal  $\Delta\psi_m$  hyperpolarization, provided the respiratory chain is not impaired and the inner membrane remains impermeable to  $H^+$  (Nicholls, 1982).

While raising interesting questions on the functionality of MELC mitochondria, the above experiments demonstrate that cell pretreatment with antimycin totally abolishes the contribution of  $\Delta\psi_m$  to  $AR_{exp}$ , even when this contribution tends to become very high (as it is in oligomycin-treated cells). In fact,  $AR_{ANT}$  and the corresponding  $\Delta\psi_p$  TPP<sup>+</sup> values calculated by Eq. (2b), in no case resulted in significantly different values in oligomycin-treated cells as compared to the controls (about -20 mV).

### THE EFFECTS OF DCCD

Up to 50 μM, DCCD produced a highly variable, but always considerable, increase of  $AR_{exp}$ . Furthermore, when the cells were pretreated with antimycin,  $AR_{ANT}$  was enhanced by DCCD at rather constant levels ( $\approx 10$ ), which were significantly higher than those displayed by the cells treated with antimycin alone or with antimycin plus oligomycin. On the contrary, the effects of DCCD were totally abolished by cell pretreatment with FCCP, showing a relevant difference in the sensitivity of these effects to protonophores or respiratory chain inhibitors.

The dose response of  $AR_{exp}$  to DCCD in cells pretreated, or not, with antimycin is given in Fig. 2. In the absence of antimycin, the function has bell-

**Table 5.** Comparison between TPP<sup>+</sup> and patch-clamp measurements of  $\Delta\psi_p$  under the standard conditions

Inoculum	$\Delta\psi_p$ (mV)				
	10 <sup>5</sup> LK	( $E_K$ ) (mV)	9 × 10 <sup>5</sup> LK	9 × 10 <sup>5</sup> HK	( $E_K$ ) (mV)
$\Delta\psi_p$ (whole-cell)	-53.7 ±9.1 (9)	-98	-20.2 ±1.6 (15)	-6.8	-10
$\Delta\psi_p$ TPP <sup>+</sup> (-61.5 log $AR_{ANT}$ )	-58.2 ±5.0 (3)	-87	-18.2 ±3.7 (8)	-1.6	-3.8

Cells were seeded at 10<sup>5</sup> or 9 × 10<sup>5</sup>/ml in complete standard medium (low K<sup>+</sup> = LK) or in the high K<sup>+</sup> medium (high K<sup>+</sup> = HK) prepared by substituting all NaCl of the standard medium with 108 mM KCl plus 24 mM KHCO<sub>3</sub> (final [K<sup>+</sup>]<sub>e</sub> = 132 mM). Antimycin A (20 μM) was added 30 min after cell seeding. After 60 min of incubation, cells were used either for patch-clamp experiments (whole-cell configuration) or processed for determination of  $AR_{exp}$ . Values of  $\Delta\psi_p$  TPP<sup>+</sup> were calculated from this ratio by means of Eq. (2b) (see Materials and Methods) and are means ± SEM of the number of experiments listed in parentheses. Whole-cell measurements refer to values recorded at the steady state of the traces and are means ± SEM of the number of determinations listed in parentheses.  $E_K$  values derive from the following parameters:  $\Delta\psi_p$  (whole cell): [K<sup>+</sup>]<sub>i</sub> = 120 mM; [K<sup>+</sup>]<sub>e</sub> = 3 mM (LK) or 80 mM (HK).  $\Delta\psi_p$  TPP<sup>+</sup>: [K<sup>+</sup>]<sub>i</sub> = 152 mM (see Table 3); [K<sup>+</sup>]<sub>e</sub> = 6 mM (LK) or 132 mM (HK) (Arcangeli et al., 1987a).



**Table 6.** Effects of ATPase inhibitors on  $AR_{exp}$  and their response to various cell pretreatments

ATPase inhibitor	Oligomycin ( $\mu\text{g}/10^7$ cells)		DCCD ( $\mu\text{M}$ )		DES ( $\mu\text{M}$ ) 50	Vanadate (mM)		Ouabain (mM) 6	
	4	40	50	200		1	5		
Pretreatment									
—	2.66 $\pm 0.21$ (21)	18.49 $\pm 2.9$ (10)	41.34 $\pm 7.5$ (7)	45.90 $\pm 10.9$ (10)	17.83 $\pm 1.8$ (6)	2.14 $\pm 0.34$ (4)	58.61 $\pm 10.2$ (5)	62.99 $\pm 11.9$ (9)	2.68 $\pm 0.24$ (11)
Antimycin	2.13 $\pm 0.35$ (8)	2.54 $\pm 0.7$ (5)	2.84 $\pm 0.6$ (3)	10.39 $\pm 1.9$ (6)	18.79 $\pm 3.2$ (4)	1.52 $\pm 0.19$ (5)	17.03 $\pm 3.3$ (5)	34.82 $\pm 3.8$ (7)	—
FCCP	2.29 $\pm 0.03$ (2)	2.73 $\pm 0.01$ (2)	—	3.13 $\pm 0.55$ (5)	—	—	—	20.03 $\pm 3.0$ (6)	—
DES	—	—	—	5.58 $\pm 0.04$ (2)	—	—	—	—	—
OUAB + FCCP	—	—	—	2.91 $\pm 0.90$ (3)	—	—	—	6.03 $\pm 0.70$ (3)	—

Cells were incubated in complete standard medium at  $9 \times 10^5/\text{ml}$ . ATPase inhibitors were added after 1 hr of incubation at the final concentrations indicated. Two min after this addition, cells were labeled with  $^3\text{H-TTP}^+$  and  $AR_{exp}$  was determined as described in Materials and Methods. Pretreatments were performed by adding the various substances at the following times before the addition of ATPase inhibitors: antimycin (20  $\mu\text{M}$ ), ouabain (OUAB) (6 mM), 30 min; FCCP (1  $\mu\text{M}$ ), DES (50  $\mu\text{M}$ ), 5 min. Values are means  $\pm$  SEM of the number of experiments listed in parentheses.

shaped kinetics, with a peak at about 75  $\mu\text{M}$ , followed by a rapid fall. On the contrary, in antimycin-pretreated cells,  $AR_{exp}$  shows a parabolic kinetics and an asymptotic approach to about 18 between 100 and 200  $\mu\text{M}$ . On the whole, we conclude that DCCD causes two distinct effects on  $AR_{exp}$  of MELC: and (i) an antimycin- and FCCP-sensitive stimulation, which reflects the  $\Delta\psi_m$  enhancement produced by the block of  $F_0-F_1$ -ATPase and is abolished by the dose excess; (ii) an antimycin-insensitive but FCCP-sensitive stimulation, which accounts for the whole  $\text{TPP}^+$  accumulation beyond 150  $\mu\text{M}$  DCCD. Since we know from oligomycin experiments that the  $\Delta\psi_m$  hyperpolarization produced by the block of the  $F_0-F_1$ -ATPase is entirely prevented by the antimycin pretreatment, these data suggest that DCCD causes an intracellular  $\text{TPP}^+$  accumulation that cannot be attributed to mitochondrial hyperpolarization. On the other hand, being prevented by FCCP, this accumulation seems sustained by a proton disequilibrium across, or within, cellular membranes.

#### THE EFFECTS OF DES

While the effects of DCCD on  $AR_{exp}$  are shared by another widely used inhibitor of  $\text{H}^+$ -ATPase, such as N-ethylmaleimide (*not shown*), a totally different behavior is displayed by DES. In fact, as shown in Table 6, up to concentrations of 50  $\mu\text{M}$ , that block the  $\text{H}^+$  extrusion rate to the same extent as DCCD (*see* Table 2), DES has no significant effect on  $AR_{exp}$ , either in the absence or in the presence of antimycin.

#### THE DES-ANTAGONIZATION OF DCCD EFFECTS

Pretreatment with DES antagonized the DCCD-dependent increase of  $AR_{exp}$ , so that  $AR_{DES+DCCD}$  was much lower than  $AR_{DCCD}$ , and even significantly lower than  $AR_{ANT+DCCD}$ . In other words, DES did not only prevent the  $\Delta\psi_m$  hyperpolarization, but it also substantially antagonized the effects of DCCD on the extramitochondrial parameters affecting  $AR_{exp}$ .

These data deserve special attention in view of the peculiarity of DES interaction with the  $F_0-F_1$ -ATPase (McEnery & Pedersen, 1986; McEnery, Hüllihen & Pedersen, 1989). Like oligomycin and DCCD, DES blocks the proton-translocating channel of this ATPase, but, unlike the other two inhibitors, it opens an alternative proton pathway at the  $F_0$ -lipid interface. This bypasses the effects of the block of  $F_0$ , operated either by DES itself or by other  $F_0$  inhibitors, such as DCCD. Under these conditions, DES should fail to induce any  $\Delta\psi_m$  hyperpolarization but should also antagonize the hyperpolarizing activity of DCCD. By analogy, assuming that the block of the proton channel of the plasma membrane  $\text{H}^+$ -ATPase is implicated in the  $\Delta\psi_p$  hyperpolarization sustained by DCCD (recall Fig. 1 and Table 1), the bypass of this block via the alternative pathway created by DES would explain both the lack of effects of the latter on  $AR_{exp}$  and its antagonization of the extramitochondrial activity of DCCD. This interpretation would imply that the hyperpolarizing effects of DCCD involve functional changes of the target pump antagonized by a  $\text{H}^+$

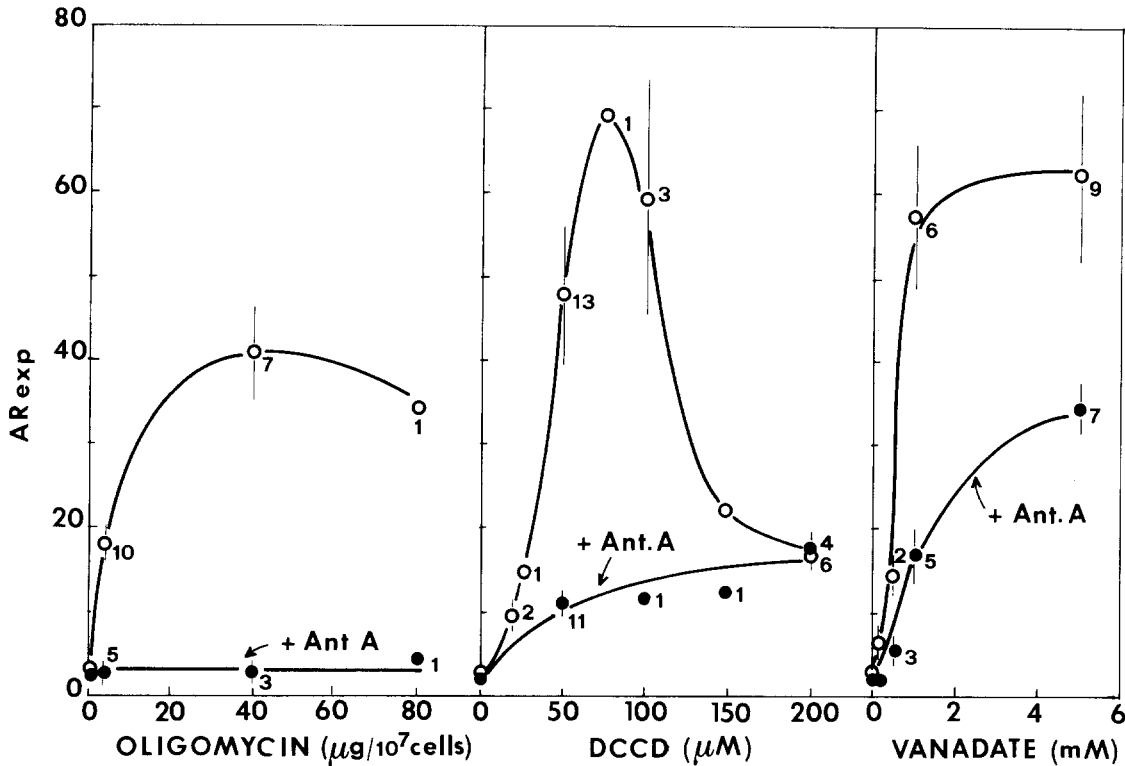


Fig. 2. Dose response of  $AR_{exp}$  to ATPase inhibitors in control and antimycin-pretreated cells. Experimental conditions were the same as reported in the legend to Table 6. Values are means  $\pm$  SEM of the number of experiments indicated by the numbers next to the symbols.

bypass at the boundaries of the pump itself. In this light, the reversal of DCCD effects by FCCP could mimic the DES dissipation of the proton gradient within, or in the vicinity of, the  $H^+$  pump.

#### THE EFFECTS OF VANADATE AND THEIR ANTAGONIZATION BY OUABAIN

As illustrated in Table 6 and Fig. 2, vanadate (1–5 mM) enhanced  $AR_{exp}$  of the controls up to about 60 and  $AR_{ANT}$  up to asymptotic values as high as 34. This leads to the conclusion that, like DCCD, this inhibitor caused a marked  $\Delta\psi_m$  hyperpolarization, displaying an oligomycin-similar activity on mitochondria. This similarity is also suggested by the vanadate effects on cellular ATP (see Table 4). On the other hand, the bulk of the  $AR_{exp}$  response to vanadate, unlike that to DCCD, is equally insensitive to FCCP as to antimycin, reflecting the difference in the target affected by the two ATPase inhibitors. However, the following effects of ouabain indicate that, also in the case of vanadate, functional changes of the target pump are implicated in this response. In fact, 6 mM ouabain, that blocked the

$Na^+, K^+$ -ATPase to the same extent as 5 mM vanadate (see Table 3), had no effect on  $AR_{exp}$  but drastically prevented the FCCP-insensitive effects of vanadate. It is worth noting here that vanadate and ouabain block the  $Na^+, K^+$ -ATPase with different mechanisms (Beaugè, 1988; Hootman & Ernst, 1988), which, paradoxically, involve antagonistic effects on the so-called "occluded state" of the pump. The latter is an intermediate state characterized by the sequestration of  $K^+$  within the pump molecule during transport (Post, Hegyvary & Kume, 1972; Glynn & Richards, 1982; Forbush, 1987). In fact vanadate stabilizes this state, while ouabain prevents it (Glynn & Richards, 1982). Thus, it is tempting to speculate that the stabilization of the occluded state is necessary, if not sufficient, to confer on the pump the properties that induce the increase in extramitochondrial uptake of  $TPP^+$ .

To summarize, data presented in Table 6 lead to the conclusion that both DCCD and vanadate determine a marked increase of  $AR_{exp}$  that includes two contributions: first, an antimycin- and FCCP-sensitive component due to the increase in  $\Delta\psi_m$ ; and second, the bulk of which is antimycin and FCCP insensitive is apparently sustained by modifications

**Table 7.** Effects of *HK* medium on  $AR_{\text{exp}}$  and estimate of  $\text{TPP}^+$  binding either in the absence or in the presence of DCCD and vanadate

	$AR_{HK+ANT}$	$HK\text{-TPP}_{\text{we}}^+$ (%)	$AR_{HK\text{-FCCP}}$	$AR_{ANT}^a$	$\text{TPP}_{\text{we}}^+$ (%)	$C_B^{HK}$ ( $AR_{HK+ANT}^{-1}$ ) <sup>b</sup>	$C\Delta\psi_p^c$	$C_B^d$	$C_B/C\Delta\psi_p^*$
ATPase inhibitor									
—	1.24 $\pm 0.4$ (4)	118 $\pm 40$	—	2.13	112 $\pm 12$	0.24	1.13	0	—
DCCD 50 $\mu\text{M}$	3.09 $\pm 0.25$ (6)	106 $\pm 9$	1.77 $\pm 0.29$ (2)	10.39	99.1 $\pm 16$	2.09	3.24	6.15	1.9
Vanadate 5 mM	3.34 $\pm 0.26$ (3)	97.1 $\pm 13$	—	34.8	112 $\pm 4.3$	2.34	10.4	23.49	2.2

<sup>a</sup> Taken from Table 6.

<sup>b</sup> Calculated from Eq. (1d).

<sup>c</sup> Calculated from the  $\Delta\psi_p$  values, measured in patch clamp (Table 1 and Fig. 1), according to the formula:  $\Delta\psi_p = -61.5 \log(1 + C\Delta\psi_p^*)$ , or  $C\Delta\psi_p^* = 10^{\Delta\psi_p / -61.5} - 1$ .

<sup>d</sup> Calculated according to Eq. (1c) by the formula:  $C_B = AR_{ANT} - (1 + C\Delta\psi_p)$ .

Cells were incubated at  $9 \times 10^5/\text{ml}$  either in complete standard or in *HK* medium containing 132 mM  $\text{K}^+$  ( $E_K = -3.9$  mV). ATPase inhibitors were added after 1 hr of incubation at the final concentrations indicated. Two min after this addition, cells were labeled with  $^3\text{H}\text{-TPP}^+$  and  $AR_{\text{exp}}$  was determined as described in Materials and Methods. Pretreatments with antimycin A (20  $\mu\text{M}$ ) or FCCP (1  $\mu\text{M}$ ) were performed by adding these substances 30 or 5 min, respectively, before the addition of the ATPase inhibitors. Water extraction of  $\text{TPP}^+$  taken up by the cells was carried out as described in Materials and Methods; values are reported as percentages of water-extractable  $\text{TPP}^+$  on the total  $\text{TPP}^+$  taken up either in *HK* or in standard medium ( $HK\text{-TPP}_{\text{we}}^+$  and  $\text{TPP}_{\text{we}}^+$ , respectively). Values are means  $\pm$  SEM of the number of experiments listed in parentheses. For explanation see text.

of the target pumps that are conferred by some inhibitors and antagonized by others.

#### THE EFFECTS OF HIGH $[\text{K}^+]_e$ ON $AR_{\text{exp}}$ IN THE ABSENCE OR IN THE PRESENCE OF ATPASE INHIBITORS AND THE ESTIMATE OF $\text{TPP}^+$ BINDING

Patch-clamp experiments demonstrated that the  $\Delta\psi_p$  hyperpolarization produced by DCCD and vanadate, being abolished in *HK* media, is sustained by the increase in  $\text{K}^+$  conductance. Incubation in these media was then used to discriminate the extramitochondrial effects of these inhibitors due to variations of either the  $\text{K}^+$  conductance, or  $\text{TPP}^+$  binding, calculated according to Eq. (1d). As shown in Table 7, incubation in *HK* medium brought  $AR_{ANT+HK}$  close to 1 ( $\Delta\psi_p \approx 0$ ), while it did not lower this parameter below 3.0 in DCCD- or vanadate-treated cells. When compared to  $AR_{ANT}$  values taken from Table 6, these data allow the following conclusions: (i) the bulk of the extramitochondrial effects of DCCD and vanadate (revealed by  $AR_{ANT}$ ) are  $[\text{K}^+]_e$  dependent, reflecting the increase in the  $\text{K}^+$  conductance. In the light of information gathered in the preceding section, such increase should be regarded now as the consequence of the functional modifications of the target pumps induced by the ATPase inhibitors. (ii) Both DCCD and vanadate determine a substantial increase in  $\text{TPP}^+$  binding in *HK* medium ( $C_B^{HK}$ ). In the case of DCCD, this effect is again abolished by FCCP (compare the effects of DCCD

on  $AR_{HK+ANT}$  with those on  $AR_{HK+FCCP}$ ;  $P < 0.05$ ), suggesting that the  $\text{K}^+$  conductance and the  $\text{TPP}^+$  binding are increased through a similar mechanism reversed by the protonophore.

Electrophysiological measurements of  $\Delta\psi_p$ , performed in standard medium, were then utilized to calculate the parameter  $C\Delta\psi_p^*$ , i.e., the expected contribution of  $\Delta\psi_p$  to  $AR_{\text{exp}}$  in this medium. This allowed us to estimate the expected contribution to  $AR_{\text{exp}}$  of  $\text{TPP}^+$  binding in physiological media ( $C_B$ ) in the absence or in the presence of DCCD and vanadate. As shown, this parameter, that is null in the controls, is increased by the inhibitors to values that are proportional to  $C\Delta\psi_p^*$ , so that  $C_B/C\Delta\psi_p^*$  is roughly constant and equal to  $C_B^{HK}$  ( $\approx 2$ ). In other words, DCCD and vanadate apparently modify to the same extent the processes underlying the  $\text{TPP}^+$  binding either in the standard or in the *HK* medium, while the absolute amount of bound  $\text{TPP}^+$  increases proportionally to the  $[\text{TPP}^+]_i$  attained within the cell as a function of the resting potential. This trend agrees with that described for the bound-surface density of  $\text{TPP}^+$  in artificial phospholipid vesicles, which increases proportionally to the probe concentration up to an electrostatic saturation at 600  $\mu\text{M}$  (Flewelling & Hubbell, 1986a).

Finally, Table 7 shows that the amount of  $\text{TPP}^+$  recoverable from the cells by means of a freezing-thawing extraction in water ( $\text{TPP}_{\text{we}}^+$ ) represents the totality of  $\text{TPP}^+$  taken up by the cells in all the conditions tested, either in the absence or in the presence of DCCD and vanadate. This finding rules

out any hydrophobic sequestration of the probe within cellular components and fits in with the  $\text{TPP}^+$  adsorption at the boundaries of lipid bilayers as a function of their electrostatic potential profile. In this light one should conclude that DCCD and vanadate change this profile in MELC plasma membrane.

#### CONCLUDING REMARKS

Patch-clamp and  $\text{TPP}^+$ -data presented in this paper converge to show that DCCD and vanadate, although acting on different targets, increase the  $\text{K}^+$  conductance across the plasma membrane of leukemia cells. This increase requires the direct and specific interaction of these inhibitors with the target ATPases, while it is antagonized by inhibitors that block the same pumps with different mechanisms (DES or ouabain). A totally speculative explanation of this finding could be that, upon interaction with the effective inhibitor, the ion-translocating moiety of the pump is transformed into a  $\text{K}^+$ -permeable channel. In this context it should be noted that a mutant in the plasma membrane  $\text{H}^+$ -ATPase gene of *Saccharomyces cerevisiae*, with a reduced  $\text{H}^+$ -ATPase activity, when examined at the single-channel level, exhibited the activation of a  $\text{K}^+$  channel that was hardly distinguishable from the  $\text{H}^+$ -ATPase itself (Ramirez et al., 1989).

The other important information to emerge from this work derived from the integration of patch-clamp and  $\text{TPP}^+$  data, showing that both DCCD and vanadate alter the complex set of membrane electrostatic parameters reflected by the  $\text{TPP}^+$  binding (see Introduction and comments on Table 7). It is worth recalling here that the internal energy profile of lipid bilayers can be changed by molecules modifying their dipole potential (Hladky, 1979; Kleijn et al., 1983; Flewelling & Hubbell, 1986b). We propose here that DCCD and vanadate may alter substantially the membrane energy profile by varying the internal dipole potential and/or the double-layer surface potential. This effect, undetectable by means of electrophysiological measurements, could: (i) alter the  $\text{TPP}^+$  adsorption at the membrane water interfaces, accounting for the increase of the binding demonstrated in Table 7 and, hence, of the alterations of  $AR_{\text{exp}}$  reported in Table 6; and (ii) be monitored by potential-sensitive proteins spanning the membrane. One of these molecules could be a voltage-sensitive  $\text{K}^+$  channel either pre-existing in a silent configuration or newly formed across the pump upon interaction with specific inhibitors. Alternatively, changes of the electrostatic potential profile sustained by the ATPase inhibitors could drastically enhance the  $\text{Ca}^{2+}$  concentration at the boundaries

of the plasma membrane, thus activating calcium-sensitive  $\text{K}^+$  conductance. This can occur even at prohibitive concentrations of this ion in the bulk cellular water. The crucial role of the electrostatic surface potential in determining the  $\text{Ca}^{2+}$  concentration necessary at the binding sites for triggering exocytosis has been clearly established by McLaughlin and Whitaker (1988) in sea urchin eggs.

However, the fact that, in the case of the  $\text{Na}^+, \text{K}^+$ -ATPase, the increase in  $\text{K}^+$  conductance is activated by vanadate and prevented by ouabain suggests that putative changes of the electrostatic potential profile of the membrane are somehow related to the creation of the occluded state of this pump. This would favor the hypothesis that occlusion of cations within the ion-translocating moiety of the pumps could locally modify the membrane internal energy profile by interacting, for instance, with the membrane dipole potential (Flewelling & Hubbell, 1986a). This interpretation could also apply to the effects of DCCD on the  $\text{H}^+$ -ATPase of the plasma membrane, if DCCD creates the occlusion of  $\text{H}^+$  within or in the vicinity of the  $\text{H}^+$  pump. The bypass of this occlusion by FCCP and DES would thus explain the reversal of DCCD effects operated by these substances on either the  $\text{TPP}^+$  binding or the  $\text{K}^+$  conductance.

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#### References

- Al-Awqati, Q. 1986. Proton-translocating ATPases. *Annu. Rev. Cell Biol.* **2**:179–199
- Amzel, L.M., Pedersen, L. 1983. Proton ATPases: Structure and mechanism. *Annu. Rev. Biochem.* **52**:801–824
- Andersen, O.S., Feldberg, S., Nakadomari, H., Levy, S., McLaughlin, S. 1978. Electrostatic interactions among hydrophobic ions in lipid bilayer membranes. *Biophys. J.* **21**:35–70
- Arcangeli, A., Del Bene, M.R., Olivotto, M., Wanke, E. 1989b. Ion channels in cancer cells. *In: Membrane Technology*. Vol. 64, pp. 65–79, Raven, New York

- Arcangeli, A., Del Bene, M.R., Poli, R., Ricupero, L., Olivotto, M. 1989a. Mutual contact of murine erythroleukemia cells activates depolarizing cation channels, whereas contact with extracellular substrata activates depolarizing  $\text{Ca}^{2+}$ -dependent  $\text{K}^+$  channels. *J. Cell. Physiol.* **139**:1–8
- Arcangeli, A., Olivotto, M. 1986. Plasma membrane potential of murine erythroleukemia cells: Approach to measurement and evidence for cell density-dependence. *J. Cell. Physiol.* **127**:17–27
- Arcangeli, A., Ricupero, L., Olivotto, M. 1987a. Commitment to differentiation of murine erythroleukemia cells involves a modulated plasma membrane depolarization through  $\text{Ca}^{2+}$ -activated  $\text{K}^+$  channels. *J. Cell. Physiol.* **132**:387–400
- Arcangeli, A., Wanke, E., Olivotto, M., Camagni, S., Ferroni, A. 1987b. Three types of ion channels are present on the plasma membrane of Friend erythroleukemia cells. *Biochem. Biophys. Res. Commun.* **146**:1450–1457
- Ashcroft, F.M., Harrison, D.E., Ashcroft, S.J.H. 1984. Glucose induces closure of single potassium channels in isolated rat pancreatic  $\beta$ -cells. *Nature* **312**:446–448
- Azzone, G.F., Pietrobon, D., Zoratti, M. 1984. Determination of the proton electrochemical gradient across biological membranes. *Curr. Top. Bioenerg.* **13**:1–77
- Bakker, E.P. 1982. Membrane potential in a potassium transport-negative mutant of *Escherichia coli* K-12. The distribution of rubidium in the presence of valinomycin indicates a higher potential than that of the tetraphenylphosphonium cation. *Biochim. Biophys. Acta* **681**:474–483
- Baroni, M., Arcangeli, A., Di Pietro, L., Alberghina, L. 1988. Plasma membrane potential of budding yeast is modulated by glucose and by inhibitors of  $\text{H}^+$ -ATPase and of  $\text{K}^+$  channels. *Yeasts* **4**:S34
- Beaugé, L. 1988. Inhibition of translocation reaction by Vanadate. *Methods Enzymol.* **156**:251–267
- Benos, D.J. 1982. Amiloride: A molecular probe of sodium transport in tissues and cells. *Am. J. Physiol.* **242**:C131–C145
- Cantley, L.C., Resh, M.D., Guidotti, G. 1978. Vanadate inhibits the red cell ( $\text{Na}^+$ ,  $\text{K}^+$ )ATPase from the cytoplasmic side. *Nature* **272**:552–554
- De La Peña, P., Barros, F., Gascon, S., Ramos, S., Lazo P.S. 1982. The electrochemical proton gradient of *Saccharomyces*. *Eur. J. Biochem.* **123**:447–453
- Felber, S.M., Brand, M.D. 1982. Factors determining the plasma-membrane potential of human lymphocytes. *Biochem. J.* **204**:577–585
- Flewelling, R.F., Hubbell, W.L. 1986a. Hydrophobic ion interaction with membranes. *Biophys. J.* **49**:531–540
- Flewelling, R.F., Hubbell, W.L. 1986b. The membrane dipole potential in a total membrane potential model. *Biophys. J.* **49**:541–552
- Forbush, B., III. 1987. Rapid release of  $^{42}\text{K}$  and  $^{86}\text{Rb}$  from an occluded state of the Na,K-pump in the presence of ATP or ADP. *J. Biol. Chem.* **262**:11104–11115
- Futai, M., Noumi, T., Maeda, M. 1989. ATP synthase ( $\text{H}^+$ -ATPase): Results by combined biochemical and molecular biological approaches. *Annu. Rev. Biochem.* **58**:111–136
- Glynn, I.M., Karlsh, S.J.D. 1975. The sodium pump. *Annu. Rev. Physiol.* **37**:13–55
- Glynn, I.M., Richards, D.E. 1982. Occlusion of rubidium ions by the sodium-potassium pump: Its implications for the mechanism of potassium transport. *J. Physiol.* **330**:17–43
- Goffeau, A., Slayman, C.W. 1981. The proton-translocating ATPase of fungal plasma membrane. *Biochim. Biophys. Acta* **636**:197–223
- Hamill, O.P. 1983. Potassium and chloride channels in red blood cells. In: Single-channel Recording. B. Sackmann and E. Neher, editors. pp. 451–471, Plenum, New York
- Hamill, O.P., Marty, A., Neher, E., Sakmann, B., Sigworth, F. 1981. Improved patch-clamp techniques for high resolution current recording from cells and cell-free membrane patches. *Pfluegers Arch.* **391**:85–100
- Hladky, S.B. 1979. The carrier mechanism. *Curr. Top. Membr. Transp.* **12**:53–164
- Hootman, S.R., Ernst, S.A. 1988. Estimation of  $\text{Na}^+$ ,  $\text{K}^+$ -pump numbers and turnover in intact cells with [ $^3\text{H}$ ]ouabain. *Methods Enzymol.* **156**:213–229
- Hoppe, J., Sebald, W. 1984. The proton conducting  $F_0$ -part of bacterial ATP synthases. *Biochim. Biophys. Acta* **768**:1–27
- Jordan, P.C. 1983. Electrostatic modeling of ion pores. II. Effects attributable to the membrane dipole potential. *Biophys. J.* **41**:189–195
- Ketterer, B., Neumcke, B., Läuger, P. 1971. Transport mechanism of hydrophobic ions through lipid bilayer membranes. *J. Membrane Biol.* **5**:225–245
- Kleijn, W.B., Bruner, L.J., Midland, M.M., Wisniewski, S. 1983. Hydrophobic ion probe studies of membrane dipole potentials. *Biochim. Biophys. Acta* **727**:357–366
- Lichtshtein, D., Mullikin-Kilpatrick, D., Blume, A.J. 1982. Hyperpolarization of neuroblastoma-glioma hybrid NG 108-15 by vanadium ions. *Proc. Natl. Acad. Sci. USA* **79**:4202–4206
- McEnery, M.W., Hüllihen, J., Pedersen, P.L. 1989.  $F_0$  “proton channel” of rat liver mitochondria. *J. Biol. Chem.* **264**:12029–12036
- McEnery, M.W., Pedersen, P.L. 1986. Diethylstilbestrol. A novel  $F_0$ -directed probe of the mitochondrial proton ATP-ase. *J. Biol. Chem.* **261**:1745–1752
- McLaughlin, S. 1977. Electrostatic potentials at membrane-solution interfaces. *Curr. Top. Membr. Transp.* **25**:71–144
- McLaughlin, S. 1989. The electrostatic properties of membranes. *Annu. Rev. Biophys. Chem.* **18**:113–136
- McLaughlin, S., Whitaker, M. 1988. Cations that alter surface potentials of lipid bilayers increase the calcium requirement for exocytosis in sea urchin eggs. *J. Physiol.* **396**:189–204
- Mitchell, P. 1961. Coupling of phosphorylation to electron and hydrogen transfer by a chemi-osmotic type of mechanism. *Nature* **191**:144–148
- Neumcke, B., Läuger, P. 1969. Nonlinear electrical effects in lipid bilayer membranes. II. Integration of the generalized Nernst-Planck equations. *Biophys. J.* **9**:1160–1170
- Nicholls, D.G. 1982. Bioenergetics. An introduction to the Chemiosmotic Theory. Academic, London
- Pedersen, P.L., Carafoli, E. 1987a. Ion motive ATPases. I. Ubiquity, properties and significance to cell function. *Trends Biochem. Sci.* **12**:146–150
- Pedersen, P.L., Carafoli, E. 1987b. Ion motive ATPases. II. Energy coupling and work output. *Trends Biochem. Sci.* **12**:186–189
- Penefsky, H.S. 1979. Mitochondrial ATPase. *Adv. Enzymol.* **49**:223–280
- Pogolotti, A.L., Santi, D.V. 1982. High pressure liquid chromatography-ultraviolet analysis of intracellular nucleotides. *Anal. Biochem.* **126**:335–345
- Post, R.L., Hegyvary, C., Kume, S. 1972. Activation by adenosin triphosphate in the phosphorylation kinetics of sodium and potassium ion transport adenosine triphosphatase. *J. Biol. Chem.* **247**:6530–6540
- Ramirez, J.A., Vacata, V., McCusker, J.H., Haber, J.E., Mortimer, R.K., Owen, W.G., Lecar, H. 1989. ATP-sensitive  $\text{K}^+$  channels

- in a plasma membrane H<sup>+</sup>-ATPase mutant of the yeast *Saccharomyces cerevisiae*. *Proc. Natl. Acad. Sci. USA* **86**:7866–7870
- Ritchie, R.J. 1984. A critical assessment of the use of lipophilic cations as membrane potential probes. *Prog. Biophys. Mol. Biol.* **43**:1–32
- Rottenberg, H. 1979. The measurement of membrane potential and pH in cells, organelles and vesicles. *Methods Enzymol.* **55**:547–569
- Skou, J.C. 1988. Overview: The Na,K-pump. *Methods Enzymol.* **156**:1–25
- A. Arcangeli et al.: ATPase Inhibitors and Membrane Potential
- Swallow, C.J., Grinstein, S., Rotstein, O.D. 1988. Cytoplasmic pH regulation in macrophages by an ATP-dependent and N,N'-dicyclohexylcarbodiimide-sensitive mechanism. *J. Biol. Chem.* **263**:19558–19563
- Wallick, E.J., Schwartz, A. 1988. Saturation of cardiac glycosides with Na<sup>+</sup>,K<sup>+</sup>-ATPase. *Methods Enzymol.* **156**:201–213

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